

$$\begin{cases} N + P = \delta \tau \\ N = e + v \\ e = p - r \end{cases}$$

$$\Rightarrow e + v + e + r = \delta \tau$$

- 1

$$\gamma \delta r = e + r$$

$$e = r \quad p = r$$

$$\frac{(\cancel{r} \times \gamma) + (\cancel{r} \times r)}{e} \quad (1)$$

$$\frac{N + r}{\delta} = \frac{1 \cdot r}{\delta} = r \gamma$$

- 9

$$M = \gamma \cdot \delta \text{ mol H}_2\text{O} \times \frac{\gamma_1 \cdot r \cdot r}{1 \text{ mol H}_2\text{O}} \times \frac{r \cdot r}{1 \text{ H}_2\text{O}} = \gamma_1 \cdot r \cdot r$$

- 10



$$M_{\text{mol}} = \gamma \text{ g CH}_4 \times \frac{1 \text{ mol CH}_4}{17 \text{ g CH}_4} = \gamma \cdot r$$

- 11

$$M_{\text{g Fe}} = \gamma \cdot r \cdot e \times \frac{1 \text{ mol Fe}}{\gamma_1 \cdot r \cdot r} \times \frac{\delta \tau}{1 \text{ mol Fe}} = \gamma \cdot r$$

- 12